

# Calculating moist enthalpy from usual meteorological measurements.

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**Abstract:** This short article shows how to compute the moist enthalpy from usual meteorological measurements of dry temperature, air pressure and relative humidity. The result is used to add a plot of moist air enthalpy to the other near-live graphs shown by meteoLCD, the meteorological station of the LCD, Diekirch, Luxembourg

## 1. Sensible heat of dry air

The sensible heat of dry air is defined as  $H_a = C_p * T$  [ref. 3] with  $C_p$  usually taken as 1.005 when  $H_a$  is given in [kJ/kg] and temperature  $T$  in [°C]. Here we will use for  $C_p$  the following expression, valid for temperatures higher than 0 °C and lower than 60 °C, as given by PADFIELD [ref.2]

$$H_a = 1.007 * T - 0.026 \quad 0 \text{ °C} < T < 60 \text{ °C} \quad [\text{eq.1}]$$

## 2. Heat content of water vapor at temperature T

The heat content of water vapor is the sum of the latent heat of vaporization and the sensible heat of water vapor:

$$H_v = q * (L + 1.84 * T) \quad [\text{ref. TET}] \quad [\text{eq.2}]$$

Where  $L$  = heat of vaporization = 2501 kJ/kg at 0°C  
and  $1.84 * T$  = sensible heat of water vapor in kJ/kg

The sensible heat term of eq.3 ( $1.84 * T$ ) is very often considered negligible and omitted.

### **Note:**

$L$  is a function of temperature, becoming slightly smaller with increasing  $T$ ; for values between 0°C and 50°C one can use the linear interpolation  $L(T) = 2502 - 2.378 * T$  computed by the author from a table with enthalpy values given by YHCHEN [ref.4]: The linear fit is excellent with  $R^2 = 0.9998$ .  
Combining eq.2 with  $L(T)$  gives:

$$H_v = q*(2502 - 0.538*T) \quad \text{with } H_v \text{ in kJ/kg and } T \text{ in } ^\circ\text{C} \quad [\text{eq.3}]$$

### 3. Total enthalpy of moist air

Total enthalpy of moist air is the sum of  $H_a$  and  $H_v$ :

$$H = H_a + H_v = (1.007*T - 0.026) + q*(2502 - 0.538*T) \quad [\text{eq.4}]$$

with  $H$  in kJ/kg,  $T$  in  $^\circ\text{C}$  and specific humidity  $q$  in kg/kg

The problem with this formula is that the **specific** humidity  $q$  is usually not measured by a standard meteorological equipment which commonly measures **relative** humidity.

### 4. Finding $q$ from measured dry bulb temperature, relative humidity and atmospheric pressure

PIELKE [ref.3] and the AOMIP website [ref.1] give the following formula for the specific humidity  $q$ :

$$q = \frac{0.622 * e_a}{p_a - 0.378 * e_a} \quad [\text{eq.5}]$$

where  $e_a$  = vapor pressure in [Pa] and  $p_a$  = atmospheric pressure in [Pa].

Attention:  $p_a$  is the true air pressure, not the barometric pressure reduced to sea level!

Dividing numerator and denominator by  $e_a$  gives:

$$q = \frac{0.622}{\frac{p_a}{e_a} - 0.378} \quad [\text{eq.6}]$$

Relative humidity is the fraction of water vapor pressure to saturated water vapor pressure, usually multiplied by 100 to give a percent value:

$$\text{RH} = 100 * e_a / e_{\text{sat}} \rightarrow e_a = \text{RH} / 100 * e_{\text{sat}}$$

There are many different formulas relating  $e_{\text{sat}}$  to temperature. We will use the expression given in AOMIP [ref.1] and valid up to  $40^\circ\text{C}$ :

$$e_{\text{sat}} = 10^{\left[ \frac{0.7859 + 0.03477 * T}{1 + 0.00412 * T} + 2 \right]} \quad [\text{eq.7}]$$

with saturated water vapor pressure  $e_{\text{sat}}$  in [Pa] and temperature  $T$  in °C.

Equations 4, 6 and 7 contain only  $T$ ,  $RH$  and  $p_a$ , which are parameters measured by practically every standard weatherstation. Together they can be used to calculate the enthalpy of moist air by a single (albeit unwieldy) formula:

$$H = (1.007 * T - 0.026) + (2502 - 0.538 * T) * \frac{0.622}{\frac{\frac{P_a}{RH} * 10}{100} \frac{0.7859 + 0.03477 * T}{1 + 0.00412 * T} - 0.378} - 0.378$$

[eq.8]

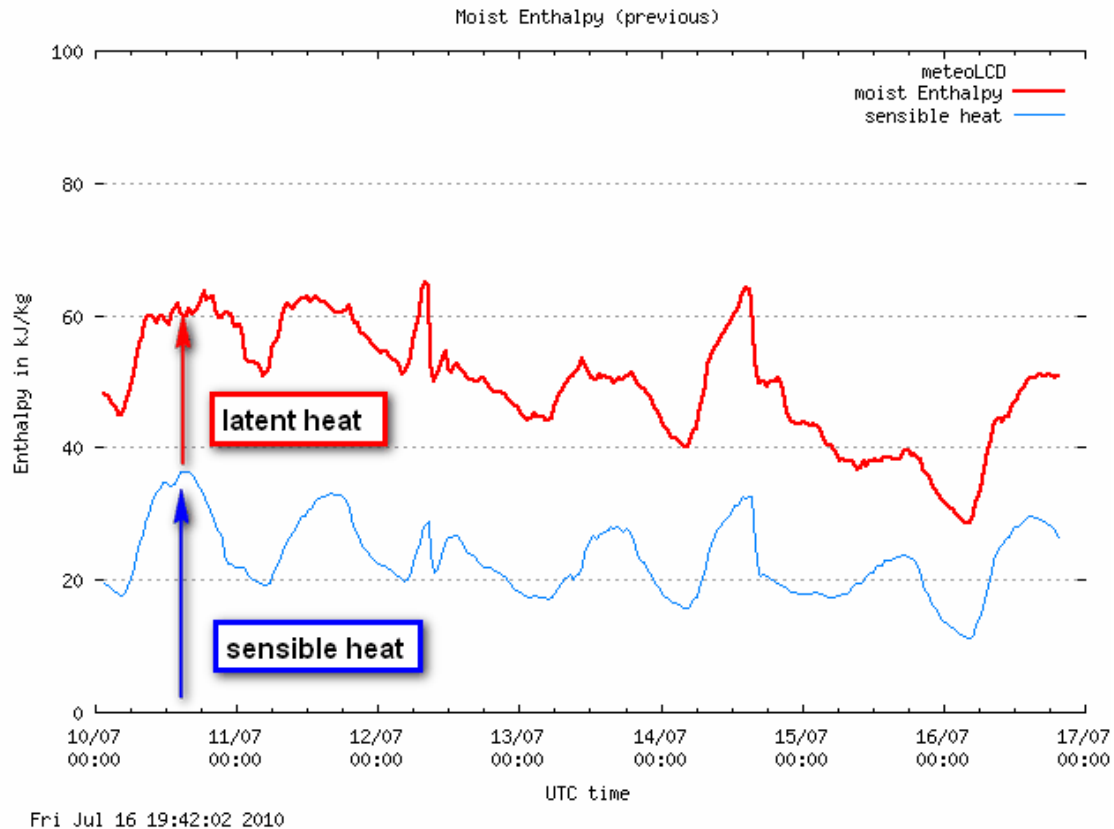
This expression is valid for temperatures  $0^\circ\text{C} < T < 40^\circ\text{C}$ . Units:  $H$ [kJ/kg],  $T$ [°C],  $p_a$ [Pa]

## 5. A practical example

The author has used eq.8 in GNUPLOT to display near-live plots of the moist enthalpy at meteoLCD, Diekirch, Luxembourg (see [http://meteo.lcd.lu/today\\_01.html](http://meteo.lcd.lu/today_01.html)). The following figure shows the situation for the week from 10 to 16<sup>th</sup> July 2010. Sensible heat is shown by the blue bottom curve; the difference between the upper red curve (= moist enthalpy) and the blue curve corresponds to the latent heat.

Technisolve Software has a website with an online moist air calculator, which is very handy for a quick validation check of individual values:

<http://www.coolit.co.za/airstate/airmoistobject.htm>



## References

- [1] AOMIP: Atmospheric Forcing Data – Humidity  
[http://efdl.cims.nyu.edu/project\\_aomip/forcing\\_data/atmosphere/humidity.html](http://efdl.cims.nyu.edu/project_aomip/forcing_data/atmosphere/humidity.html)
- [2] PADFIELD, Tim: Conservation Physics  
<http://www.conservaionphysics.org/atmcalc/atmoclc1.php>
- [3] PIELKE, Roger, Sr., WOLTER, Klaus: The July 2005 Denver Heat Wave: How unusual was it ?. National Weather Digest, vol.31, no. 1, July 2007  
<http://pielkeclimatesciencesci.files.wordpress.com/2009/10/r-313.pdf>
- [4] TET (The Engineering Toolbox)  
[http://www.engineeringtoolbox.com/enthalpy-moist-air-d\\_683.html](http://www.engineeringtoolbox.com/enthalpy-moist-air-d_683.html)
- [5] YHCHEEN: Calculation of Enthalpy Changes  
[www.ntut.edu.tw/~yhchen1/Chap.%2023.pdf](http://www.ntut.edu.tw/~yhchen1/Chap.%2023.pdf)